METALLURGICAL STRUCTURE MODIFICATION OF UO₂ PELLET DURING SINTERING - EXPERIENCE AT NFC, HYDERABAD, INDIA

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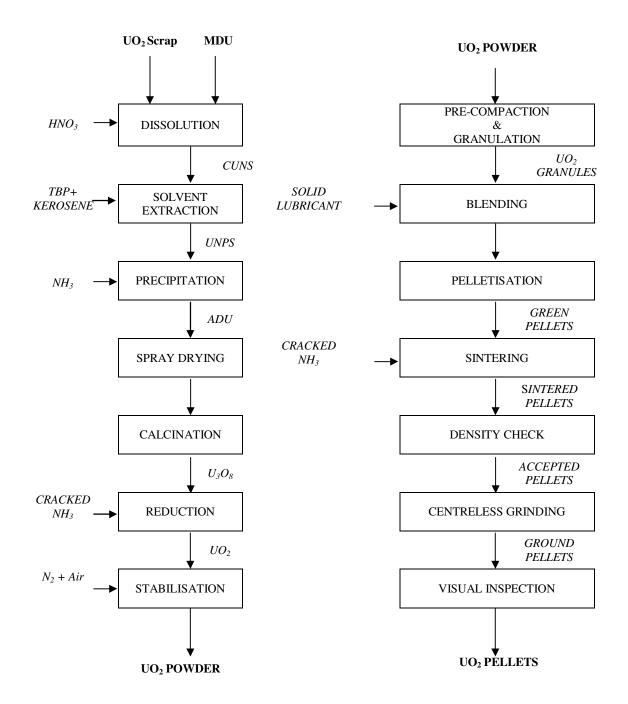
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ABSTRACT: Nuclear Fuel Complex (NFC), Department of Atomic Energy (DAE) produces UO₂ fuel pellets by powder compaction, high temperature sintering followed by centreless wet grinding method from the stabilized UO₂ powder generated through ADU-route. Enhancement of fuel burn up of the Indian PHWRs becomes very important in order to effectively utilize the fuel to the maximum extent inside the reactor. Burn up is mainly limited by increased fission gas release from the fuel during reactor operation. Without introducing much change in the design, rate of release of fission gas can be reduced through enlargement of UO₂ grain size. In Powder Metallurgical (PM) route of fuel fabrication, trials were taken by doping various oxide powder additives like TiO2, Al2O3, SiO2, Nb2O5 and Cr2O3. The dopant normally goes into the solid solution of parent matrix during sintering at 1700°C and thus enhance the rate of diffusion. Aliovalant dopant can alter the defect chemistry of the parent material either by creating vacancy or interstitial. It is apparently understood that the combination of above mechanisms are responsible for structural modification of UO₂. Hence selection of dopant remains largely empirical. It has been observed at NFC Hyderabad that the Cr₂O₃ is the most suitable for achieving average UO₂ grain size of about 70 micron and 98%TD of the sintered pellet. The paper discusses about the various experimental trials, sintered densities, metallographic examination, effect of different quantities, analysis and result obtained thereof:

Introduction

Ceramic uranium dioxide (UO₂) in the form of cylindrical sintered pellets are widely used as nuclear fuel material for power generation all over the world. Quality specification is very stringent in terms of fuel density and integrity for the natural UO₂ pellet used in Indian PHWR's. Effective utilization of nuclear fuel inside the reactor is of prime importance w.r.t fuel burn up and to avoid frequent refueling in the reactor. Fission product gases like Cesium (Cs), Tellurium (Te), Iodine (I) etc generates from the fuel lattice during reactor operation. They are corrosive in nature and once release from the fuel matrix, come into contact with Zr-alloy clad and corrode them by stress corrosion cracking (SCC). Fuel assembly needs to be removed from the reactor prematurely with partial utilization of fissile materials. Retention of fission gases may be possible by the modification of the microstructure of pellets. An enlarged UO₂ grain is expected to retain more fission gases over a larger period of time inside the reactor. Introduction of optimized quantity of suitable sintering aid can promote grain growth and density during high temperature sintering in reducing cracked ammonia atmosphere.

The process flow sheet for the production of UO₂ powder and pellets at NFC:-



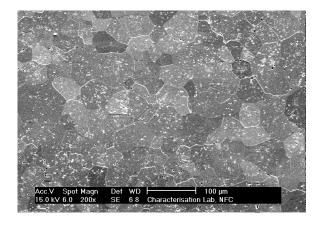
1. Experimental method

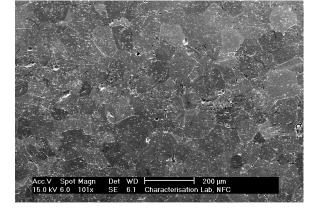
Five categories of sintering aids in the form of ceramic oxide powder additives viz. Nb₂O₅, Al₂O₃, TiO₂, Cr₂O₃ and SiO₂ were selected for the trial with ADU-ex UO₂ powder. The median size of the UO₂ powder agglomerate is about 12-14 micron. The initial O/U ratio of the powder was 2.045 and BET specific surface area was 2.9-3.1 m²/g. The additives are

incorporated in UO2 granules which were produced in pre-compaction of as received UO2 powder at about 80-100 MPa pressure to make the powder free flowable for final compaction. These powders were added very less quantity so are called as powder dopants. They were mixed thoroughly with UO₂ powder granule in varying quantities of 0.2wt%, 0.3wt%, 0.4wt%, and 0.5wt% for each category of blend. Quantities are chosen in view that the optimum results should come with the lowest quantity dopant addition. Total twenty numbers of blends were prepared with the above combination taking the same precipitation lot of ADU-ex UO₂ powder in order to avoid variability in UO₂ powder characteristics. Homogenization of additive mixing were ensured by preparing master blend at first with highest quantity of dopant and then making dilution as per the various percentages required for the trials. Powder blends were compacted in double acting die compaction with admixed solid lubricant (0.3wt%) at a pressure of about 275-300 MPa to obtain green density in the range of 52-54%TD as in the regular production. Pellets were sintered in high temperature (1700°C) continuous pusher type sintering furnace having six different temperature zones along with regular production of conventional nuclear grade UO₂ pellets without any dopants for comparison. Soaking period of 9-10 hrs was ensured at temperature 1700°C. Cracked ammonia atmosphere were maintained inside the furnace as reducing atmosphere. All the other parameters of powder compaction and sintering were maintained same as in the regular commercial production. The sintered pellet samples were selected randomly for sintered densities and metallographic examination. Pellet samples were cut, polished and etched with 10% H₂SO₄ + 90% H₂O₂ solution to reveal the microstructure in SEM. Repeated trials were carried out with changing the lot number of ADU-ex UO₂ powder keeping the other parameters in same condition. Chemical compositional analysis of sintered pellets were carried out for confirming homogenization, presence of additives and calculation for neutron absorption factor in the reactor.

2. Microstructures

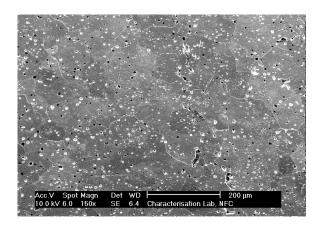
SEM photo micrographs for the pellets having two different quantities lowest 0.2 wt% and the highest 0.5 wt% of dopant in each category are analyzed here for grain size measurement.





0.2 wt% Cr₂O₃

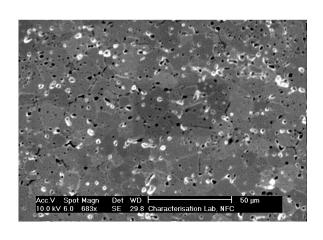
 $0.5 \text{ wt}\% \text{ Cr}_2\text{O}_3$

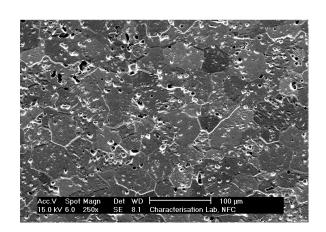


Acc V Spot Magn | Det WD | 200 µm | 50 kV 6.0 121x | SE 6.8 Characterisation Lab, NFC

0.2 wt% TiO2

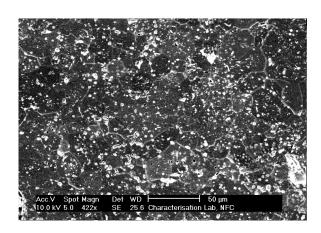
0.5 wt% TiO2

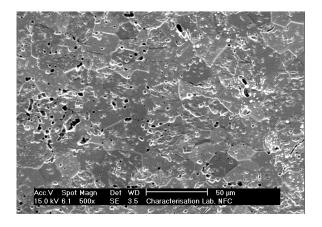




0.2 wt% Nb₂O₅

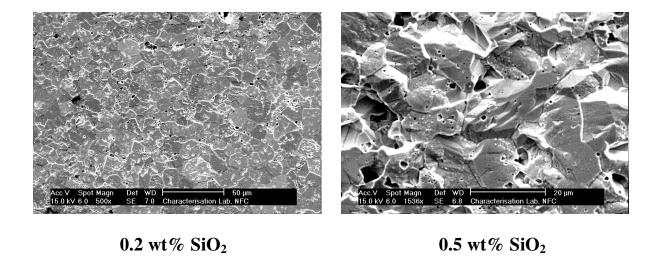
0.5 wt% Nb₂O₅





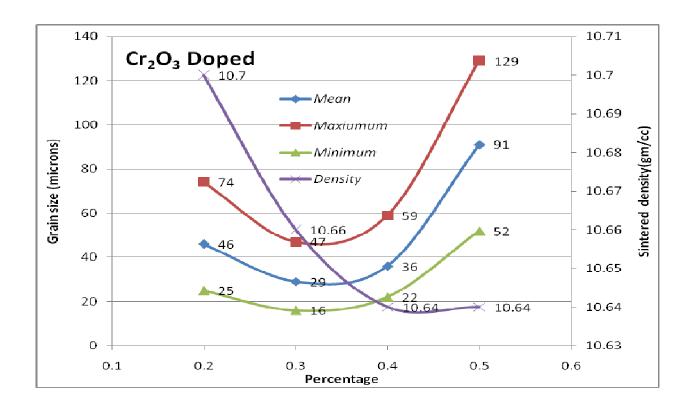
0.2 wt% Al₂O₃

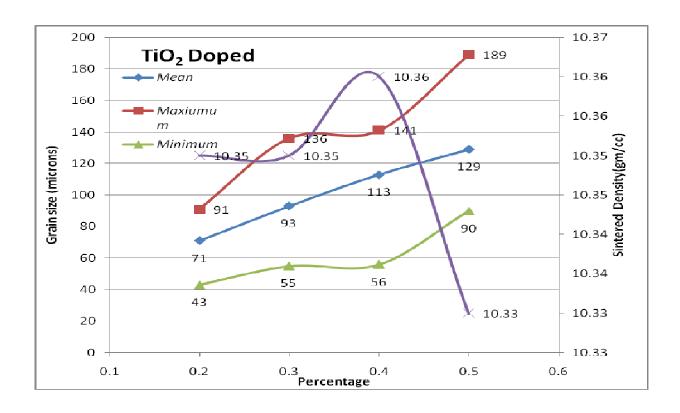
0.5 wt% Al₂O₃

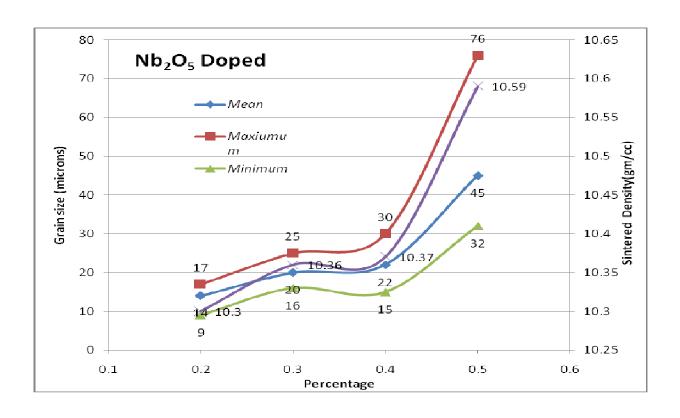


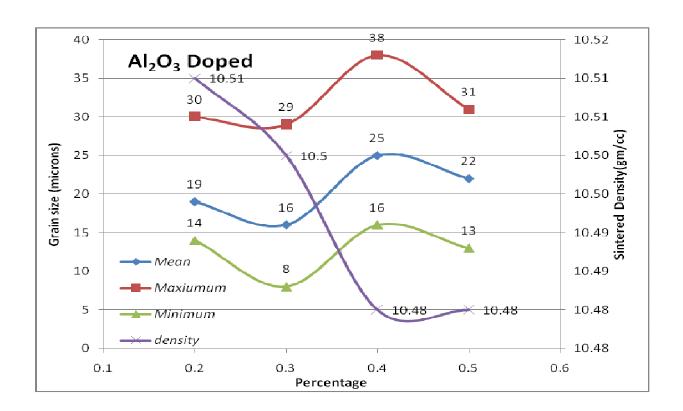
3. Graphical representations on effect of dopants

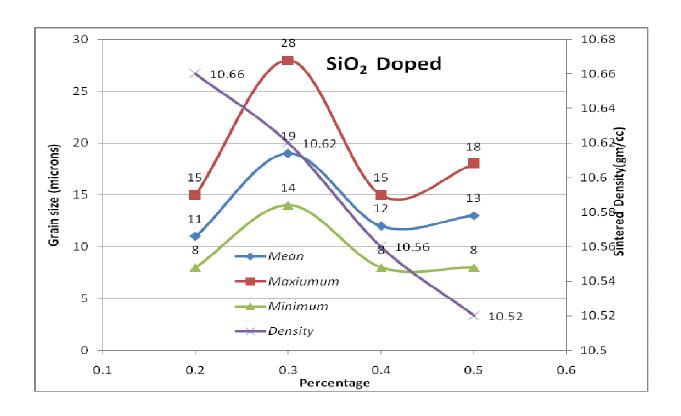
Graphs are prepared taking maximum, minimum and mean grain size values with the quantity of dopant and superimposing the sintered density values on same plot for every dopant.

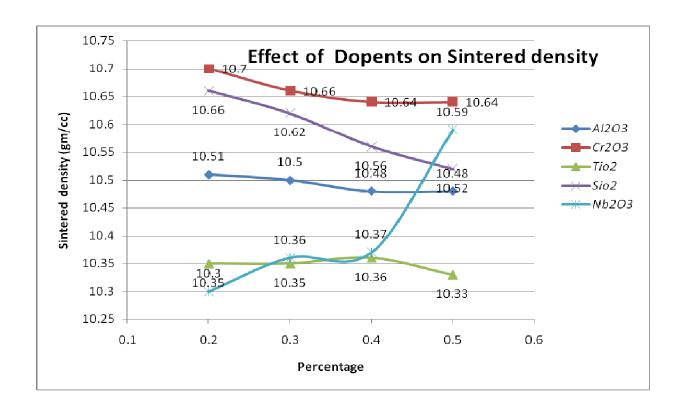


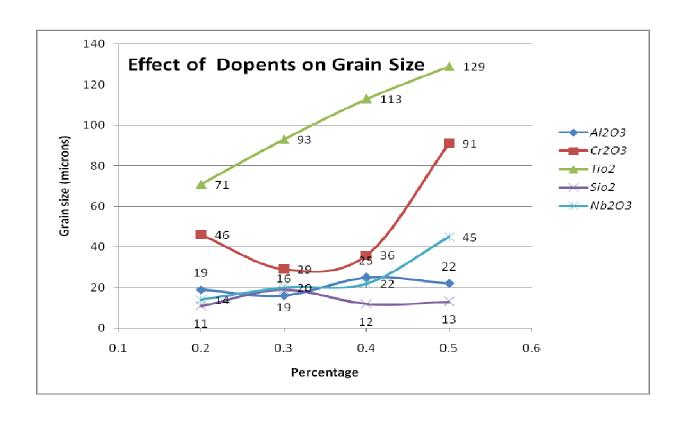












4. Results analysis and conclusions

In case of Cr₂O₃ doped pellets, the average grain size is increasing with the quantity of Cr₂O₃ above 0.3% but the sintered density of the pellet decreases with increasing percentage but maintaining higher values. For TiO₂, the grain size is increasing steadily with the percentage increases but the sintered density values are low compare to other additives and hence not acceptable. In case of Nb₂O₅, the grain size is remarkably enlarged from 22µ to 45µ when the quantity increases from 0.4% to 0.5% but the density increases modestly from 10.37 to 10.59 gm/cc. The optimum quantity for Al₂O₃ is 0.4%. The density decreases with increasing quantity. In case of SiO_2 , the graph looks like Al_2O_3 except the optimum quantity here is 0.3%. The density decreases when quantity increased but maintaining reasonably higher density values. Any dopant acts as an impurity in the reactor especially in PHWR is concerned w.r.t neutron absorption point of view. Therefore reasonably good results at lowest quantity addition is always expected during reactor operation. In summary, the grain size effectively increases with TiO₂ and Cr₂O₃ at lower quantity addition. Cr₂O₃ and SiO₂ are giving reasonably good sintered densities among all other dopants at lower percentage. It appears from the trials that the dopant Cr₂O₃ with 0.2 wt% is the most suitable choice for increasing the grain size and sintered density both in UO₂. The experimental condition needs to be checked number of times in order to obtain reproducible results.

5. Discussion and future scope

The high temperature sintering of reactor grade ADU-ex UO₂ pellet in reducing atmosphere gives grain size of about 10-25µ without addition of any sintering aid. The UO₂ grain size of about 70-80µ is achieved with addition of 0.2 wt% Cr₂O₃ as an external doping agent. Some additives like Al₂O₃, SiO₂ might appear in the form of a liquid glassy phases along the grain boundaries at high temperature and thus increases the diffusion rate by the mechanism of liquid phase sintering. Other soluble aliovalent dopant probably goes into the solid solution, creates vacancies or interstitials depending upon the oxidation states of dopant cations Nb^{4+, 5+}, Cr^{3+, 4+}, Ti^{3+, 4+} etc and thus change the defect chemistry of UO₂ lattice during solid state sintering [1][2]. Any defects inside UO₂ lattice lower the activation energy and enhance the diffusion rate. In case of TiO₂, some of Ti⁴⁺ ions substitute U⁴⁺ ions but few Ti³⁺ ions may form interstitial solid solution with UO₂ [3]. Diffusion of slow moving U⁴⁺ ion is activated by interstitial mechanism of diffusion. Segregation of dopant can alter the structure and composition of surfaces thereby influence the thermodynamic factors by modifying the interfaces and grain boundary energy during sintering [4]. It has been observed that the solubility limit for Cr₂O₃ dopant in UO₂ is about 0.07% at 1700°C. The excess dopant above solubility limit generally acts as inclusion and thus inhibit grain growth [1] by pinning the grain boundaries through Zener mechanism. The grain growth at 0.2wt% Cr₂O₃ may indicate formation of low melting eutectic (Cr-CrO) at temperature of about 1500-1660°C in Cr-Cr₂O₃-UO₂ system in presence of SiO₂ [1]. It is therefore assumed the dissolution and precipitation of UO₂ takes place through liquid secondary phase [1]. Although, exact mechanism of grain growth in UO₂ is not clear yet, but it is apparently understood that the combination of one with the others are mostly responsible for microstructure modification of UO₂. Additional information about the effect of dopant in reactor operating

condition like irradiation behavior, coefficient of reactivity, neutron absorption, thermal conductivity, induced radioactivity etc are necessary before final conclusion for the selection of dopant to promote grain growth of UO₂.

6. Acknowledgement

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7. References

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